



L-Proline-catalyzed five-component domino reaction leading to multifunctionalized 1,2,3,4-tetrahydropyridines

Huan-Feng Jiang*, Jing-Hao Li, Zheng-Wang Chen

School of Chemistry and Chemical Engineering, South China University of Technology, Guangzhou 510640, PR China

ARTICLE INFO

Article history:

Received 17 July 2010

Received in revised form 28 September 2010

Accepted 15 October 2010

Available online 11 November 2010

ABSTRACT

Multifunctionalized 1,2,3,4-tetrahydropyridines are concisely synthesized in good yields via L-proline-catalyzed or L-proline/FeCl₃-cocatalyzed one-pot multicomponent reactions (MCRs). The MCRs involve a domino hydroamination/prins reaction/Mannich-type reaction/intramolecular dehydration–cyclization process. The molecular structure of **5baa**, one of multifunctionalized 1,2,3,4-tetrahydropyridines, was confirmed by single-crystal X-ray diffraction.

© 2010 Elsevier Ltd. All rights reserved.

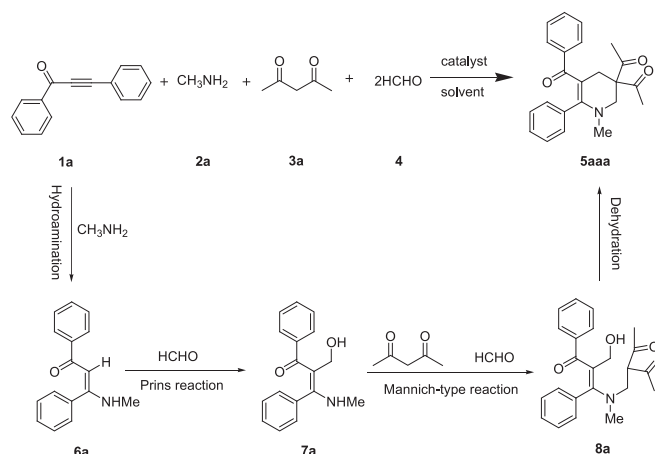
1. Introduction

Multicomponent reactions (MCRs) have emerged as a powerful tool in synthetic organic chemistry. Compared with conventional methods, MCRs exhibit high levels of efficiency and diversity, as they offer rapid and convergent construction of complex molecular skeletons from common starting materials in one-pot, and these reactions are environmentally friendly and proceed with excellent chemoselectivities.^{1,2} Therefore, designing novel MCRs for constructing valuable compounds remains the interest of synthetic chemists.

Tetrahydropyridines are one of the fundamental heterocycles, which have been the subject of intense research for their outstanding biological properties and wide range of applications to pharmaceutical compounds and synthetic intermediates.³ Although there are some reports for the synthesis of tetrahydropyridines, these methods are not always satisfactory with respect to ease of operation, yield, applicability and the regioselectivity of the products.⁴ Besides, very few efficient methods of formation 1,2,3,4-tetrahydropyridines are described in the literatures.^{4d,5}

Over the past few years, we have developed a series of heterocyclic compounds, such as tetrahydropyrimidines,⁶ 3,6-dihydro-2H-1,3-oxazines,⁷ 1,3-oxazine-6-ones,⁸ 1,4-dihydropyridines,⁹ 1,2,3,4-tetrahydropyridines^{5g} through the MCR methods. During our ongoing research on the synthesis of other nitrogen-containing heterocycles, we found a novel multicomponent coupling strategy to synthesize multisubstituted 1,2,3,4-tetrahydropyridine derivatives. Our approach could briefly comprise four sequential processes (Scheme 1): (1) two-component hydroamination;¹⁰ (2)

two-component prins reaction;¹¹ (3) three-component Mannich-type reaction;¹² (4) intramolecular dehydration–cyclization.

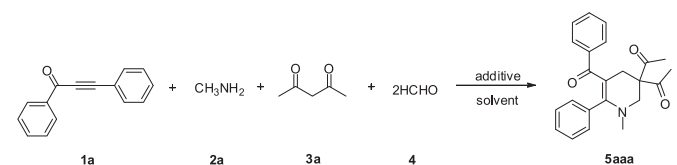


Scheme 1. The domino processes leading to multisubstituted 1,2,3,4-tetrahydropyridine.

2. Results and discussion

On the basis of our previous work on the L-proline-catalyzed synthesis of highly functionalized multisubstituted 1,4-dihydropyridines,^{9,12} we initially examined the reaction in ethanol solvent at room temperature, the reaction did not give target product (Table 1, entry 1). To our delight, the reaction gave satisfying results in DMF at 100 °C (Table 1, entry 2). Without any catalysts, the reaction was carried out inefficiently, and the desired product was obtained in a poor yield of 33% (Table 1, entry 3). Next,

* Corresponding author. E-mail address: jianghf@scut.edu.cn (H.-F. Jiang).

Table 1
Optimization of reaction conditions^a


Entry	Additive	Solvent	Yield ^b (%)
1 ^c	L-Proline	EtOH	n.p.
2	L-Proline	DMF	87 (83)
3	None	DMF	33
4	L-Proline	EtOH	Trace
5	L-Proline	Toluene	65
6	L-Proline	CH ₃ CN	64
7	L-Proline	THF	56
8	L-Proline	DMSO	77
9	Pyridine	DMF	75
10	DABCO	DMF	77
11	Et ₃ N	DMF	46
12	NaOH	DMF	35
13	Na ₂ CO ₃	DMF	36
14	CH ₃ COONa	DMF	39
15 ^d	L-Proline	DMF	68
16 ^e	L-Proline	DMF	75

^a Reaction conditions: **1a** (1 mmol), **2a** (1 mmol), **3a** (1.5 mmol), **4** (2 mmol), and additive (0.1 mmol, 10 mol %) in solvent (3.0 mL) at 100 °C for 12 h.

^b Determined by GC. Number in parenthesis is isolated yield.

^c Room temperature.

^d 60 °C.

^e 6 h.

the solvents were evaluated. Except ethanol, other solvents such as toluene, CH₃CN, THF, and DMSO proved to be appropriate (Table 1, entries 4–8). The reaction also showed a significant dependence on bases, and L-proline proved to be superior to other bases (Table 1, entries 2 and 9–14). The lower temperature and the shortened reaction time can reduce the product yield (Table 1, entries 15 and 16).

On the basis of these results, the optimal conditions involved the following parameters: L-proline as catalyst, and DMF as solvent, with reaction temperature at 100 °C for 12 h. Under these optimized conditions, a study on the substrate scope was carried out, and the results are summarized in Table 2.

In order to determine the structures of this series of compound definitively, a single crystal of **5baa** was obtained by slow crystallization from ethyl acetate and its structure was established by single-crystal X-ray analysis (Fig. 1).

Under the above optimized conditions, the scope of this new MCR process was next examined using various readily available starting materials (Table 2). It was pleasing to find that all the reactions afforded the corresponding products in good yields. The alkynones with different substituted groups on the benzene ring had no obvious impact to the reaction. The heterocyclic alkynone afforded the corresponding products in high yields. Unfortunately, only the simple primary amines, such as methylamine and ethylamine, were suitable for the reaction. Moreover, a series of active methylene compounds were compatible for the reaction.¹³ When the reaction substrates were switched from β-diketones to malononitrile, the reactions could be carried out smoothly to give the corresponding products with higher yields (Table 2, entries 1–11).

Table 2
L-Proline-catalyzed synthesis of polysubstituted 1,2,3,4-tetrahydropyridines via the MCRs^a

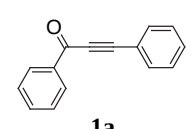
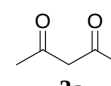
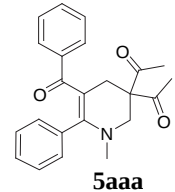
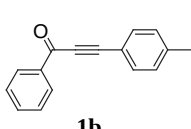
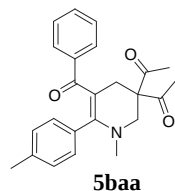
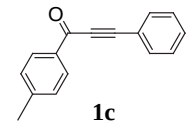
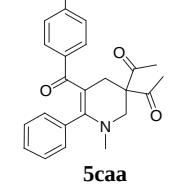
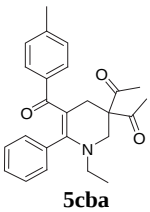
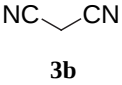
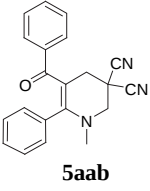
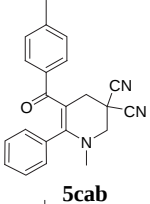
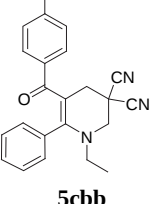
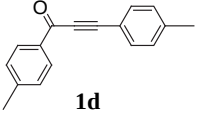
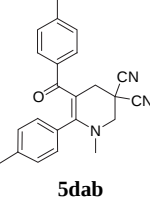
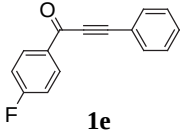
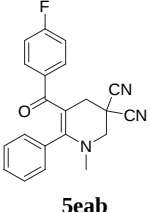
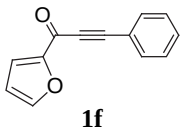
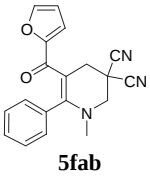
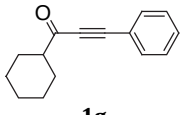
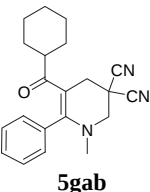
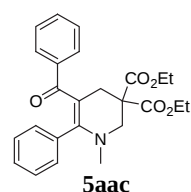
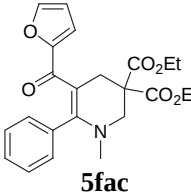
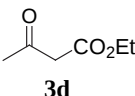
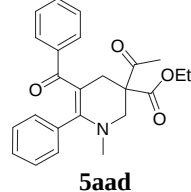
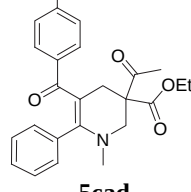
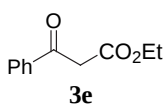
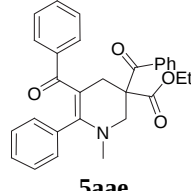

Entry	Alkynone	Amine	Active methylene compound	Product	Yield ^b (%)
1	 1a	 2a	 3a	 5aaa	83
2	 1b	2a	3a	 5baa	84
3	 1c	2a	3a	 5caa	81

Table 2 (continued)

Entry	Alkynone	Amine	Active methylene compound	Product	Yield ^b (%)
4	1c	 2b	3a	 5cba	77
5	1a	2a	 3b	 5aab	86
6	1c	2a	3b	 5cab	82
7	1c	2b	3b	 5cbb	82
8	 1d	2a	3b	 5dab	89
9	 1e	2a	3b	 5eab	78
10	 1f	2a	3b	 5fab	83
11	 1g	2a	3b	 5gab	84

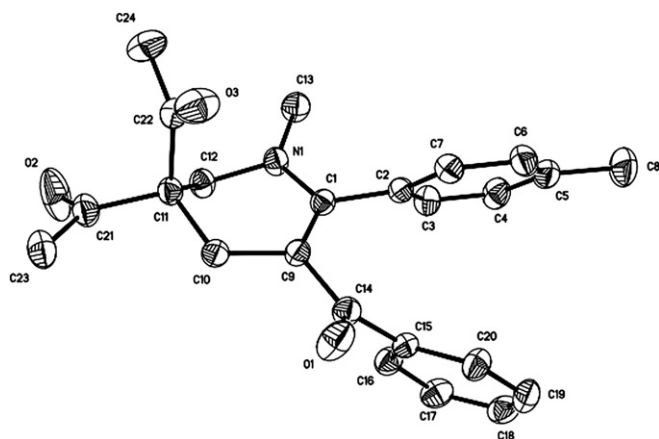
(continued on next page)

Table 2 (continued)

Entry	Alkynone	Amine	Active methylene compound	Product	Yield ^b (%)
12	1a	2a	EtOOC-CH ₂ -COOEt 3c	 5aac	68
13	1f	2a	3c	 5fac	64
14	1a	2a	 3d	 5aad	78
15	1c	2a	3d	 5cad	83
16	1a	2a	 3e	 5aae	74

^a Reaction conditions: **1** (1 mmol), **2** (1 mmol), **3** (1.5 mmol), **4** (2 mmol), and L-proline (10 mol %) in DMF (3.0 mL) at 100 °C for 12 h.

^b Isolated yield.

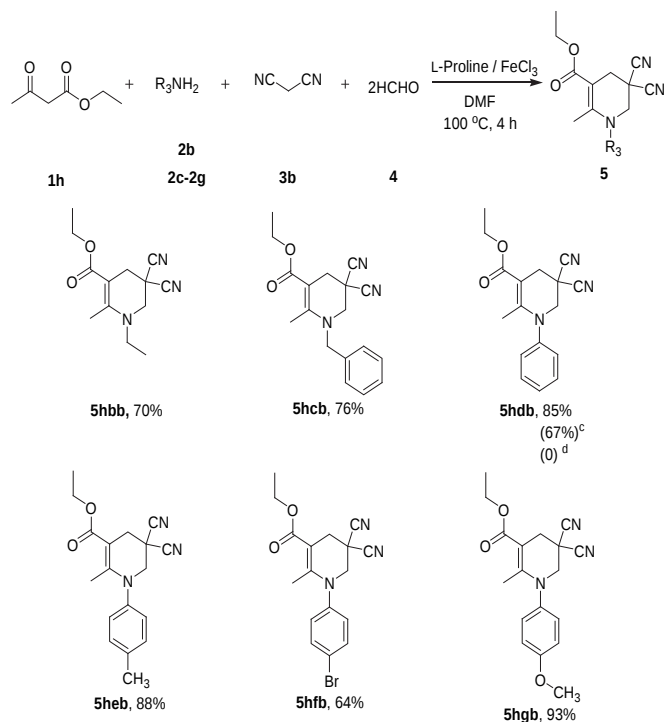
Fig. 1. X-ray structure of **5baa**.

When β -diketones were replaced with diethyl malonate, the reaction gave the products with lower yields (Table 2, entries 12–13). This meant that steric effect had much influence on the reaction.

For further exploration the scope of the MCRs, L-proline/ FeCl_3 -catalyzed system was developed. Table 2 has showed an elegant method for the synthesis of multifunctionalized 1,2,3,4-tetrahydropyridines, but the substrates were limited to the aromatic alkynones and aliphatic amines. Herein we got a series of 1,2,3,4-tetrahydropyridines from ethyl acetoacetate, amines, malononitrile, and formaldehyde via L-proline/ FeCl_3 -catalyzed five-component domino reactions (Scheme 2).^{2b,14} The reaction did not proceed without FeCl_3 , and gave a reduced yield without L-proline. Aromatic amines with electron-donating attached to the benzene rings gave higher yields than the electron-withdrawing groups (Scheme 2, **5heb**, **5hfb**, and **5hgb**).

3. Conclusions

In conclusion, we have developed a new strategy that provides an efficient synthesis of multisubstituted 1,2,3,4-tetrahydropyridines and is therefore complementary to the construction of small heterocyclic molecules in our laboratory. Our method involves mild reaction conditions, uses very simple and accessible starting materials and solvents, as well as an inexpensive and non-toxic catalyst. It also has the additional advantage of proceeding with high



Scheme 2. L-Proline/ FeCl_3 -cocatalyzed synthesis of polysubstituted 1,2,3,4-tetrahydropyridines via the MCRs^{a, b}; ^a Reaction conditions: ethyl acetoacetate (1 mmol), amines (1 mmol), malononitrile (1.5 mmol), formaldehyde (2 mmol), L-proline (10 mol %) and FeCl_3 (10 mol %) in DMF (3.0 mL) at 100 °C for 4 h ^b Isolated yield. ^c Without L-proline. ^d Without FeCl_3 .

atom economy and having water as the only side product. Further synthetic applications and studies of the mechanism of the MCRs are underway.

4. Experimental

4.1. General

All the reactions were carried out under air atmosphere in a round bottom flask equipped with magnetic stir bar. Melting points were measured with a BÜCHI B-545 melting point instrument and were uncorrected. GC–MS was obtained using electron ionization (EI). IR spectra were obtained as potassium bromide pellets or as liquid films between two potassium bromide pellets with a Bruker Vector 22 spectrometer. ^1H NMR and ^{13}C NMR spectra were recorded using a Bruker Avance 400 or 600 MHz NMR spectrometer and, respectively, referenced to 7.27 and 77.0 ppm for chloroform-*d* with TMS as internal standard. TLC was performed using commercially prepared 100–400 mesh silica gel plates (GF₂₅₄), and visualization was effected at 254 nm. All the other chemicals were purchased from Aldrich Chemicals.

4.2. General procedure for the synthesis of multifunctionalized 1,2,3,4-tetrahydropyridines (5aaa–5aae)

To a mixture of alkynone **1** (1 mmol) and amine **2** (1 mmol), 3 mL DMF was added successively. The mixture was stirred at room temperature for 30 min. Subsequently, active methylene compound **3** (1.5 mmol), 38% formaldehyde **4** (160 mg, 2 mmol), and L-proline (11.5 mg, 0.1 mmol) were added, and the stirring was continued in a round bottom flask at 100 °C for another 12 h. The progress of the reaction was followed by TLC. After cooling, the reaction was diluted with water and extracted with diethyl ether (15 mL×3). The

ether layer was then washed with saturated salt water and dried over anhydrous MgSO_4 . Solvent was removed in vacuo, and the crude product was purified by preparative TLC (GF₂₅₄) with petroleum ether/ethyl acetate (10:1 to 4:1) as eluent to afford the desired products **5aaa–5aae**.

4.2.1. 1,1'-(5-Benzoyl-1-methyl-6-phenyl-1,2,3,4-tetrahydropyridine-3,3-diyl)diethanone (5aaa). Yellow viscous oil. IR (KBr): 2926, 1741, 1688, 1545, 1366, 1244, 1049, 761 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ =7.10–7.08 (m, 2H), 7.02–6.89 (m, 8H), 3.72 (s, 2H), 3.26 (s, 2H), 2.68 (s, 3H), 2.27 (s, 6H). ^{13}C NMR (100 MHz, CDCl_3): δ =204.3, 196.5, 158.3, 142.2, 134.8, 130.1, 129.1, 128.5, 128.2, 127.8, 127.1, 105.9, 65.5, 53.8, 41.0, 29.9, 26.7. MS (EI) m/z : 361, 319, 118, 105, 77, 43, 28. Anal. Calcd for $\text{C}_{23}\text{H}_{23}\text{NO}_3$: C, 76.43; H, 6.41; N, 3.88. Found: C, 76.61; H, 6.33; N, 3.84.

4.2.2. 1,1'-(5-Benzoyl-1-methyl-6-p-tolyl-1,2,3,4-tetrahydropyridine-3,3-diyl)diethanone (5baa). Yellow crystal, mp: 177–178 °C. IR (KBr): 2921, 1743, 1693, 1547, 1367, 1243, 1048, 755 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ =7.05–6.90 (m, 5H), 6.74 (s, 4H), 3.69 (s, 2H), 3.23 (s, 2H), 2.69 (s, 3H), 2.25 (s, 6H), 2.10 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3): δ =204.4, 196.6, 158.7, 142.4, 138.7, 131.9, 130.1, 128.7, 128.4, 128.1, 127.0, 105.7, 65.5, 53.7, 41.0, 29.8, 26.7, 21.0. MS (EI) m/z : 375, 333, 332, 105, 77, 32, 28. Anal. Calcd for $\text{C}_{24}\text{H}_{25}\text{NO}_3$: C, 76.77; H, 6.71; N, 3.73. Found: C, 76.91; H, 6.64; N, 3.77.

4.2.3. 1,1'-(1-Methyl-5-(4-methylbenzoyl)-6-phenyl-1,2,3,4-tetrahydropyridine-3,3-diyl)diethanone (5caa). Yellow viscous oil. IR (KBr): 2929, 1742, 1688, 1534, 1357, 1242, 1048, 753 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ =7.03–6.95 (m, 5H), 6.88 (d, 2H, J =8.0 Hz), 6.74 (d, 2H, J =8.0 Hz), 3.68 (s, 2H), 3.20 (s, 2H), 2.65 (s, 3H), 2.24 (s, 6H), 2.14 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3): δ =204.4, 196.5, 157.4, 142.4, 139.4, 135.1, 130.1, 129.1, 128.4, 127.8, 127.7, 105.9, 65.5, 53.7, 41.0, 29.8, 26.7, 21.2. MS (EI) m/z : 375, 333, 332, 119, 118, 91, 28. Anal. Calcd for $\text{C}_{24}\text{H}_{25}\text{NO}_3$: C, 76.77; H, 6.71; N, 3.73. Found: C, 76.92; H, 6.63; N, 3.80.

4.2.4. 1,1'-(1-Ethyl-5-(4-methylbenzoyl)-6-phenyl-1,2,3,4-tetrahydropyridine-3,3-diyl)diethanone (5cba). Yellow viscous oil. IR (KBr): 2928, 1743, 1699, 1545, 1359, 1224, 1075, 755 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ =7.01 (d, 2H, J =8.0 Hz), 6.99–6.88 (m, 5H), 6.74 (d, 2H, J =8.0 Hz), 3.68 (s, 2H), 3.18 (s, 2H), 2.92–2.86 (m, 2H), 2.25 (s, 6H), 2.14 (s, 3H), 1.01 (t, 3H, J =6.8 Hz). ^{13}C NMR (100 MHz, CDCl_3): δ =204.7, 196.9, 157.0, 142.7, 139.4, 135.3, 129.8, 128.6, 128.3, 127.8, 127.7, 105.9, 64.9, 50.7, 47.0, 30.3, 26.8, 21.2, 14.0. MS (EI) m/z : 389, 347, 346, 119, 91, 43. Anal. Calcd for $\text{C}_{25}\text{H}_{27}\text{NO}_3$: C, 77.09; H, 6.99; N, 3.60. Found: C, 77.26; H, 6.93; N, 3.67.

4.2.5. 5-Benzoyl-1-methyl-6-phenyl-1,2-dihydropyridine-3,3(4H)-dicarbonitrile (5aab). White crystal, mp: 225–226 °C. IR (KBr): 2926, 2351, 1672, 1607, 1559, 1382, 1267, 738, 692 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ =7.25–7.23 (m, 2H), 7.11–7.00 (m, 8H), 3.93 (s, 2H), 3.43 (s, 2H), 2.86 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3): δ =195.9, 155.7, 140.3, 133.8, 129.9, 129.8, 129.2, 127.9, 127.8, 126.9, 113.6, 106.6, 56.3, 41.3, 35.0, 28.3. MS (EI) m/z : 327, 223, 222, 207, 105, 77, 44. Anal. Calcd for $\text{C}_{21}\text{H}_{17}\text{N}_3\text{O}$: C, 77.04; H, 5.23; N, 12.84. Found: C, 77.28; H, 5.17; N, 12.79.

4.2.6. 1-Methyl-5-(4-methylbenzoyl)-6-phenyl-1,2-dihydropyridine-3,3(4H)-dicarbonitrile (5cab). White crystal, mp: 186–187 °C. IR (KBr): 2926, 2370, 1687, 1609, 1570, 1385, 1271, 827, 746, 701 cm^{-1} . ^1H NMR (400 MHz, CDCl_3): δ =7.16 (d, 2H, J =8.0 Hz), 7.06–7.03 (m, 5H), 6.80 (d, 2H, J =8.0 Hz), 3.90 (s, 2H), 3.38 (s, 2H), 2.83 (s, 3H), 2.16 (s, 3H). ^{13}C NMR (100 MHz, CDCl_3): δ =196.3, 155.6, 140.9, 137.6, 134.3, 130.3, 129.5, 128.6, 128.1, 128.0, 114.1, 107.1, 56.6, 41.8, 35.4, 28.5, 21.3. MS (EI)

m/z : 341, 222, 207, 119, 91, 32, 28. Anal. Calcd for $C_{22}H_{19}N_3O$: C, 77.40; H, 5.61; N, 12.31. Found: C, 77.18; H, 5.69; N, 12.26.

4.2.7. 1-Ethyl-5-(4-methylbenzoyl)-6-phenyl-1,2-dihydro-3,3(4H)-dicarbonitrile (5cbb). White crystal, mp: 192–193 °C. IR (KBr): 2927, 2362, 1688, 1604, 1554, 1361, 1272, 751, 703 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.14 (d, 2H, J =8.0 Hz), 7.08–7.03 (m, 5H), 6.81 (d, 2H, J =8.0 Hz), 3.87 (s, 2H), 3.38 (s, 2H), 3.08–3.03 (m, 2H), 2.17 (s, 3H), 1.09 (t, 3H, J =7.2 Hz). ^{13}C NMR (100 MHz, $CDCl_3$): δ =196.5, 155.3, 140.7, 138.0, 134.6, 130.1, 129.4, 128.5, 128.1, 128.1, 114.2, 106.2, 53.3, 47.6, 35.6, 28.6, 21.3, 14.5. MS (EI) m/z : 355, 354, 237, 236, 119, 104, 91, 65, 28. Anal. Calcd for $C_{23}H_{21}N_3O$: C, 77.72; H, 5.96; N, 11.82. Found: C, 77.98; H, 5.88; N, 11.78.

4.2.8. 1-Methyl-5-(4-methylbenzoyl)-6-*p*-tolyl-1,2-dihydropyridine-3,3(4H)-dicarbonitrile (5dab). White crystal, mp: 208–209 °C. IR (KBr): 2921, 2312, 1689, 1605, 1541, 1357, 1274, 819 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.15 (d, 2H, J =8.0 Hz), 6.93 (d, 2H, J =8.0 Hz), 6.84–6.80 (m, 4H), 3.89 (s, 2H), 3.36 (s, 2H), 2.84 (s, 3H), 2.18 (s, 3H), 2.14 (s, 3H). ^{13}C NMR (100 MHz, $CDCl_3$): δ =196.4, 155.9, 140.7, 139.7, 137.7, 131.4, 130.2, 128.8, 128.6, 128.0, 114.2, 106.9, 56.6, 41.8, 35.5, 28.6, 21.3. MS (EI) m/z : 355, 236, 119, 91, 28. Anal. Calcd for $C_{23}H_{21}N_3O$: C, 77.72; H, 5.96; N, 11.82. Found: C, 77.99; H, 5.88; N, 11.86.

4.2.9. 5-(4-Fluorobenzoyl)-1-methyl-6-phenyl-1,2-dihydropyridine-3,3(4H)-dicarbonitrile (5eab). White crystal, mp: 190–191 °C. IR (KBr): 2926, 2365, 1687, 1608, 1375, 1271, 745, 701 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.27–7.23 (m, 2H), 7.14–7.05 (m, 5H), 6.70–6.66 (m, 2H), 3.93 (s, 2H), 3.42 (s, 2H), 2.87 (s, 3H). ^{13}C NMR (100 MHz, $CDCl_3$): δ =195.0, 156.5, 136.9, 134.1, 130.8, 130.7, 130.4, 130.0, 128.4, 114.3, 106.8, 56.6, 42.0, 35.3, 28.8. MS (EI) m/z : 345, 223, 222, 122, 118, 94. Anal. Calcd for $C_{21}H_{16}FN_3O$: C, 73.03; H, 4.67; N, 12.17. Found: C, 73.21; H, 4.60; N, 12.21.

4.2.10. 5-(Furan-2-carbonyl)-1-methyl-6-phenyl-1,2-dihydropyridine-3,3(4H)-dicarbonitrile (5fab). White crystal, mp: 229–230 °C. IR (KBr): 2376, 1688, 1592, 1545, 1383, 1272, 758, 702 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.22–7.19 (m, 5H), 7.09 (s, 1H), 6.57 (d, 1H, J =3.2 Hz), 6.10–6.09 (m, 1H), 3.90 (s, 2H), 3.38 (s, 2H), 2.89 (s, 3H). ^{13}C NMR (100 MHz, $CDCl_3$): δ =182.3, 155.1, 152.7, 144.0, 134.0, 129.4, 129.2, 127.8, 115.9, 113.6, 111.0, 106.1, 56.3, 41.5, 34.6, 28.2. MS (EI) m/z : 317, 222, 207, 206, 32, 28. Anal. Calcd for $C_{19}H_{15}N_3O_2$: C, 71.91; H, 4.76; N, 13.24. Found: C, 72.14; H, 4.66; N, 13.29.

4.2.11. 5-(Cyclohexanecarbonyl)-1-methyl-6-phenyl-1,2-dihydropyridine-3,3(4H)-dicarbonitrile (5gab). White crystal, mp: 163–164 °C. IR (KBr): 2926, 2361, 1698, 1616, 1547, 1378, 1260, 769, 703 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.49–7.42 (m, 3H), 7.30–7.28 (m, 2H), 3.78 (s, 2H), 3.19 (s, 2H), 2.80 (s, 3H), 1.50–0.95 (m, 9H), 0.57–0.52 (m, 2H). ^{13}C NMR (100 MHz, $CDCl_3$): δ =204.0, 155.0, 135.4, 130.1, 129.3, 129.0, 114.1, 107.6, 56.2, 47.9, 41.7, 34.8, 29.7, 29.1, 25.7, 25.6. MS (EI) m/z : 333, 251, 250, 118, 55, 28. Anal. Calcd for $C_{21}H_{23}N_3O$: C, 75.65; H, 6.95; N, 12.60. Found: C, 75.48; H, 7.12; N, 12.64.

4.2.12. 5-Benzoyl-1-methyl-6-phenyl-1,2-dihydropyridine-3,3(4H)-dicarboxylate (5aac). Yellow viscous oil. IR (KBr): 2928, 1613, 1553, 1382, 1256, 767, 706 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.17–7.15 (m, 2H), 7.03–6.94 (m, 8H), 4.24–4.19 (m, 4H), 3.75 (s, 2H), 3.17 (s, 2H), 2.66 (s, 3H), 1.25 (t, 6H, J =7.2 Hz). ^{13}C NMR (100 MHz, $CDCl_3$): δ =196.6, 169.8, 157.5, 142.4, 135.5, 130.2, 129.1, 128.6, 128.3, 127.8, 127.1, 107.3, 61.8, 54.5, 53.1, 40.9, 31.2, 14.0. MS (EI) m/z : 421, 349, 348, 249, 170, 118, 105, 77, 28. Anal. Calcd for $C_{25}H_{27}NO_5$: C, 71.24; H, 6.46; N, 3.32. Found: C, 71.36; H, 6.38; N, 3.27.

4.2.13. Diethyl 5-(furan-2-carbonyl)-1-methyl-6-phenyl-1,2-dihydropyridine-3,3(4H)-dicarboxylate (5fac). Yellow viscous oil. IR

(KBr): 2981, 1732, 1599, 1547, 1360, 1263, 754, 702 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.14–7.09 (m, 5H), 7.07 (s, 1H), 6.44 (d, 1H, J =3.2 Hz), 6.04–6.03 (m, 1H), 4.21–4.16 (m, 4H), 3.73 (s, 2H), 3.17 (s, 2H), 2.70 (s, 3H), 1.22 (t, 6H, J =7.2 Hz). ^{13}C NMR (100 MHz, $CDCl_3$): δ =182.9, 169.6, 156.9, 153.9, 143.5, 135.7, 129.6, 128.6, 127.8, 115.0, 110.9, 106.6, 61.7, 54.5, 53.1, 40.9, 30.9, 14.0. MS (EI) m/z : 411, 339, 338, 214, 118, 95, 28. Anal. Calcd for $C_{23}H_{25}NO_6$: C, 67.14; H, 6.12; N, 3.40. Found: C, 67.35; H, 5.99; N, 3.45.

4.2.14. Ethyl 3-acetyl-5-benzoyl-1-methyl-6-phenyl-1,2,3,4-tetrahydropyridine-3-carboxylate (5aad). Reddish brown oil. IR (KBr): 2984, 1710, 1605, 1547, 1364, 1230, 730, 705 cm^{-1} . 1H NMR (600 MHz, $CDCl_3$): δ =7.12–7.10 (m, 2H), 7.03–6.91 (m, 8H), 4.26–4.25 (m, 2H), 3.82–3.67 (m, 2H), 3.27–3.20 (m, 2H), 2.69 (s, 3H), 2.33 (s, 3H), 1.29 (t, 3H, J =7.2 Hz). ^{13}C NMR (150 MHz, $CDCl_3$): δ =203.3, 196.3, 170.0, 158.3, 142.4, 135.2, 129.0, 128.7, 128.1, 127.1, 106.0, 61.9, 58.5, 53.8, 40.9, 30.9, 26.3, 14.0. MS (EI) m/z : 391, 349, 348, 118, 105, 77, 28. Anal. Calcd for $C_{24}H_{25}NO_4$: C, 73.64; H, 6.44; N, 3.58. Found: C, 73.81; H, 6.33; N, 3.44.

4.2.15. Ethyl 3-acetyl-1-methyl-5-(4-methylbenzoyl)-6-phenyl-1,2,3,4-tetrahydropyridine-3-carboxylate (5cad). Reddish brown oil. IR (KBr): 2982, 1711, 1605, 1544, 1363, 1234, 822, 749, 705 cm^{-1} . 1H NMR (600 MHz, $CDCl_3$): δ =7.04 (d, 2H, J =7.8 Hz), 7.00–6.91 (m, 5H), 6.75 (d, 2H, J =7.8 Hz), 4.25–4.21 (m, 2H), 3.78–3.64 (m, 2H), 3.23–3.15 (m, 2H), 2.66 (s, 3H), 2.29 (s, 3H), 2.14 (s, 3H), 1.27 (t, 3H, J =7.2 Hz). ^{13}C NMR (150 MHz, $CDCl_3$): δ =203.4, 196.4, 170.1, 157.7, 139.4, 135.4, 128.5, 128.4, 127.8, 127.7, 106.2, 61.9, 58.6, 53.8, 40.9, 31.1, 26.4, 21.2, 14.1. MS (EI) m/z : 405, 363, 362, 170, 119, 118, 91. Anal. Calcd for $C_{25}H_{27}NO_4$: C, 74.05; H, 6.71; N, 3.45. Found: C, 73.88; H, 6.82; N, 3.47.

4.2.16. Ethyl 3,5-dibenzoyl-1-methyl-6-phenyl-1,2,3,4-tetrahydropyridine-3-carboxylate (5aee). Yellow crystal, mp: 142–143 °C. IR (KBr): 2926, 1730, 1680, 1546, 1360, 1261, 1236, 699 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.96–7.94 (m, 2H), 7.57–7.42 (m, 3H), 7.11–6.92 (m, 10H), 4.20–4.17 (m, 2H), 3.95–3.86 (m, 2H), 3.63–3.03 (m, 2H), 2.69 (s, 3H), 1.27 (t, 3H, J =7.2 Hz). ^{13}C NMR (100 MHz, $CDCl_3$): δ =196.6, 195.5, 171.2, 158.3, 142.4, 135.6, 135.4, 133.1, 129.0, 128.7, 128.6, 128.3, 127.8, 127.0, 108.3, 61.8, 56.8, 55.9, 40.8, 32.3, 13.8. MS (ESI) m/z : 453. Anal. Calcd for $C_{29}H_{27}NO_4$: C, 76.80; H, 6.00; N, 3.09. Found: C, 76.96; H, 5.51; N, 3.06.

4.3. General procedure for the synthesis of multifunctionalized 1,2,3,4-tetrahydropyridines (5hbb–5hgb)

A mixture of ethyl acetoacetate **1h** (130 mg, 1 mmol), amines **2** (1 mmol), malononitrile **3b** (66 mg, 1 mmol), 38% formaldehyde **4** (160 mg, 2 mmol), $FeCl_3$ (16 mg, 0.1 mmol), and *L*-proline (11.5 mg, 0.1 mmol) was heated in DMF (3 ml) with stirring in a round bottom flask at 100 °C for 4 h. The progress of the reaction was followed by TLC. After cooling, the reaction was diluted with water and extracted with diethyl ether (15 mL $\times$ 3). The ether layer was then washed with saturated salt water and dried over anhydrous $MgSO_4$. Solvent was removed in vacuo, and the crude product was purified by preparative TLC (GF₂₅₄) with petroleum ether/ethyl acetate (10:1 to 4:1) as eluent to afford the desired products **5hbb–5hgb**.

4.3.1. Ethyl 5,5-dicyano-1-ethyl-2-methyl-1,4,5,6-tetrahydropyridine-3-carboxylate (5hbb). Yellow viscous oil, IR_{max} (KBr): 2977, 2260, 1766, 1681, 1568, 1424, 1256, 1202, 1160, 1106 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =4.16–4.11 (m, 2H), 3.64 (s, 2H), 3.44–3.39 (m, 2H), 3.10 (s, 2H), 2.49 (s, 3H), 1.28 (t, 3H, J =7.2), 1.21 (t, 3H, J =7.2). ^{13}C NMR (100 MHz, $CDCl_3$): δ =167.1, 154.1, 114.3, 90.7, 59.7, 53.5, 46.3, 34.2, 29.3, 15.9, 14.5, 14.1. MS (EI) m/z : 247, 202, 174, 135,

42, 29, 28. Anal. Calcd for $C_{13}H_{17}N_3O_2$: C, 63.14; H, 6.93; N, 16.99. Found: C, 63.25; H, 6.85; N, 17.04.

4.3.2. Ethyl 1-benzyl-5,5-dicyano-2-methyl-1,4,5,6-tetrahydropyridine-3-carboxylate (5hcb). Yellow viscous oil. IR (KBr): 2981, 2260, 1689, 1576, 1359, 1246, 1130, 734, 699 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.41–7.20 (m, 5H), 4.60 (s, 2H), 4.19–4.14 (m, 2H), 3.59 (s, 2H), 3.15 (s, 2H), 2.56 (s, 3H), 1.30 (t, 3H, J =7.2). ^{13}C NMR (100 MHz, $CDCl_3$): δ =167.0, 154.3, 135.9, 129.2, 128.2, 126.7, 114.2, 91.8, 59.9, 54.8, 53.6, 34.2, 29.2, 16.6, 14.5. MS (EI) m/z : 309, 264, 236, 230, 158, 132, 91, 28. Anal. Calcd for $C_{18}H_{19}N_3O_2$: C, 69.88; H, 6.19; N, 13.58. Found: C, 69.71; H, 6.27; N, 13.52.

4.3.3. Ethyl 5,5-dicyano-2-methyl-1-phenyl-1,4,5,6-tetrahydropyridine-3-carboxylate (5hdb). Reddish brown oil. IR (KBr): 2981, 2260, 1684, 1575, 1355, 1246, 1138, 734, 699 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.41–7.15 (m, 5H), 4.19–4.14 (m, 2H), 3.98 (s, 2H), 3.20 (s, 2H), 2.19 (s, 3H), 1.29 (t, 3H, J =7.2 Hz). ^{13}C NMR (100 MHz, $CDCl_3$): δ =166.7, 153.1, 143.8, 129.9, 127.8, 127.4, 114.1, 93.9, 59.9, 55.8, 34.2, 29.2, 19.2, 14.4. MS (EI) m/z : 295, 266, 250, 223, 222, 144, 118, 77. Anal. Calcd for $C_{17}H_{17}N_3O_2$: C, 69.14; H, 5.80; N, 14.23. Found: C, 69.30; H, 5.72; N, 14.18.

4.3.4. Ethyl 5,5-dicyano-2-methyl-1-p-tolyl-1,4,5,6-tetrahydropyridine-3-carboxylate (5heb). Reddish brown oil. IR (KBr): 2982, 2260, 1691, 1579, 1514, 1452, 1383, 1246, 1140, 823 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.18 (d, 2H, J =8.0 Hz), 7.03 (d, 2H, J =8.0 Hz), 4.18–4.12 (m, 2H), 3.95 (s, 2H), 3.19 (s, 2H), 2.34 (s, 3H), 2.17 (s, 3H), 1.28 (t, 3H, J =7.2 Hz). ^{13}C NMR (150 MHz, $CDCl_3$): δ =166.8, 153.4, 141.3, 137.9, 130.5, 127.3, 114.2, 93.4, 59.9, 55.9, 34.3, 29.2, 21.0, 19.1, 14.4. MS (EI) m/z : 309, 264, 236, 230, 158, 132, 91, 28. Anal. Calcd for $C_{18}H_{19}N_3O_2$: C, 69.88; H, 6.19; N, 13.58. Found: C, 69.62; H, 6.25; N, 13.62.

4.3.5. Ethyl 1-(4-bromophenyl)-5,5-dicyano-2-methyl-1,4,5,6-tetrahydropyridine-3-carboxylate (5hfb). Reddish brown oil. IR (KBr): 2981, 2260, 1694, 1577, 1513, 1487, 1383, 1244, 1145, 823 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.54 (d, 2H, J =8.4 Hz), 7.06 (d, 2H, J =8.4 Hz), 4.22–4.16 (m, 2H), 3.98 (s, 2H), 3.21 (s, 2H), 2.21 (s, 3H), 1.31 (t, 3H, J =7.2 Hz). ^{13}C NMR (100 MHz, $CDCl_3$): δ =166.6, 152.3, 142.9, 133.1, 129.0, 121.4, 114.0, 95.4, 60.1, 55.7, 34.2, 29.2, 19.3, 14.4. MS (EI) m/z : 373, 328, 300, 296, 222, 220, 196, 155, 143, 75. Anal. Calcd for $C_{17}H_{16}BrN_3O_2$: C, 54.56; H, 4.31; Br, 21.35; N, 11.23. Found: C, 54.71; H, 4.26; Br, 21.38; N, 11.20.

4.3.6. Ethyl 5,5-dicyano-1-(4-methoxyphenyl)-2-methyl-1,4,5,6-tetrahydropyridine-3-carboxylate (5hgb). Reddish brown oil. IR (KBr): 2980, 2850, 2261, 1691, 1579, 1511, 1457, 1383, 1245, 1139, 831 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$): δ =7.08 (d, 2H, J =8.8 Hz), 6.89 (d, 2H, J =8.8 Hz), 4.17–4.12 (m, 2H), 3.93 (s, 2H), 3.80 (s, 3H), 3.19 (s, 2H), 2.16 (s, 3H), 1.28 (t, 3H, J =7.2 Hz). ^{13}C NMR (150 MHz, $CDCl_3$): δ =166.8, 159.0, 153.7, 136.6, 128.8, 115.0, 114.2, 92.9, 59.8, 56.1, 55.5, 34.2, 29.2, 19.0, 14.4. MS (EI) m/z : 325, 280, 252, 247, 232, 174, 148, 28. Anal. Calcd for $C_{18}H_{19}N_3O_3$: C, 66.45; H, 5.89; N, 12.91. Found: C, 66.61; H, 5.81; N, 12.85.

4.4. X-ray data

Crystal data for **5baa**. $C_{24}H_{25}NO_3$, M =375.45, monoclinic, a =6.1835 (12), b =26.950 (5) Å, c =12.243 (2) Å, U =1983.6 (7) Å³, T =293 (2) K, space group $P2_1/n$, 15,703 reflections measured, 3577 independent reflections (R_{int} =0.0506). The final wR (F^2) was 0.1799 (all data) and R_1 was 0.0541 for $I > 2s(I)$. CCDC No. 779617.

Acknowledgements

The authors thank the National Natural Science Foundation of China (Nos. 20625205, 20772034 and 20932002), National Basic

Research Program of China (973 Program) (No. 2011CB808603), the Doctoral Fund of the Ministry of Education of China (20090172110014) and Guangdong Natural Science Foundation (No. 8451064101000236) for financial support.

Supplementary data

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.tet.2010.10.041.

References and notes

- For recent reviews of multicomponent reactions, see: (a) Dömling, A. *Chem. Rev.* **2006**, *106*, 17; (b) Ramón, D. J.; Yus, M. *Angew. Chem., Int. Ed.* **2005**, *44*, 1602; (c) Touré, B. B.; Hall, D. G. *Chem. Rev.* **2009**, *109*, 4439; (d) Tietze, L. F. *Chem. Rev.* **1996**, *96*, 115.
- For recent examples on multicomponent reactions, see: (a) Brioché, J.; Masson, G.; Zhu, J. *Org. Lett.* **2010**, *12*, 1432; (b) Maiti, S.; Biswas, S.; Jana, U. J. *Org. Chem.* **2010**, *75*, 1674; (c) Scheffelaar, R.; Paravidino, M.; Mulijk, D.; Lutz, M.; Spek, A. L.; de Kanter, F. J. J.; Orru, R. V. A.; Ruijter, E. *Org. Lett.* **2009**, *11*, 125; (d) Yoo, E. J.; Park, S. H.; Lee, S. H.; Chang, S. *Org. Lett.* **2009**, *11*, 1155; (e) Dou, G.; Shi, C.; Shi, D. J. *Comb. Chem.* **2008**, *10*, 810; (f) Bonne, D.; De-khane, M.; Zhu, J. *Angew. Chem., Int. Ed.* **2007**, *46*, 2485; (g) Li, D.-M.; Song, L.-P.; Li, X.-F.; Xing, C.-H.; Peng, W.-M.; Zhu, S.-Z. *Eur. J. Org. Chem.* **2007**, 3520; (h) Siamaki, A. R.; Arndtsen, B. A. *J. Am. Chem. Soc.* **2006**, *128*, 6050; (i) Duan, X.-H.; Liu, X.-Y.; Guo, L.-N.; Liao, M.-C.; Liu, W.-M.; Liang, Y.-M. *J. Org. Chem.* **2005**, *70*, 6980.
- (a) Aridoss, G.; Amirthaganesan, S.; Jeong, Y. T. *Bioorg. Med. Chem. Lett.* **2010**, *20*, 2242; (b) Malviya, M.; Sunil Kumar, Y. C.; Mythri, R. B.; Venkateshappa, C.; Subhash, M. N.; Rangappa, K. S. *Bioorg. Med. Chem. Lett.* **2009**, *17*, 5526; (c) Ishida, J.; Hattori, K.; Yamamoto, H.; Iwashita, A.; Mihara, K.; Matsuoka, N. *Bioorg. Med. Chem. Lett.* **2005**, *15*, 4221; (d) Suleman, N. K.; Flores, J.; Tanko, J. M.; Isin, E. M.; Castagnoli, N., Jr. *Bioorg. Med. Chem.* **2008**, *16*, 8557; (e) Wimalasena, D. S.; Perera, R. P.; Heyen, B. J.; Balasooriya, I. S.; Wimalasena, K. J. *Med. Chem.* **2008**, *51*, 760; (f) Duan, P. G.; Durant, G. J.; Rho, T.; Ojo, B.; Messer, W. S., Jr. *J. Med. Chem.* **1994**, *37*, 2774; (g) Chao, Y. X.; He, B. P.; Tay, S. S. W. *J. Neuroimmunol.* **2009**, *216*, 39; (h) Roberts, E.; Sançon, J. P.; Sweeney, J. B.; Workman, J. A. *Org. Lett.* **2003**, *5*, 4775; (i) Rodinovskaya, L. A.; Shestopalov, A. M.; Chunikhin, K. S. *Tetrahedron* **2002**, *58*, 4273.
- For some examples on the synthesis of multiplysubstituted tetrahydropyridines, see: (a) Tsukamoto, H.; Kondo, Y. *Angew. Chem., Int. Ed.* **2008**, *47*, 4851; (b) Spiegel, D. A.; Schroeder, F. C.; Duvall, J. R.; Schreiber, S. L. *J. Am. Chem. Soc.* **2006**, *128*, 14766; (c) Ramachandran, P. V.; Burghardt, T. E.; Bland-Berry, L. J. *Org. Chem.* **2005**, *70*, 7911; (d) Zhu, X.-F.; Lan, J.; Kwon, O. J. *Am. Chem. Soc.* **2003**, *125*, 4716; (e) Legault, C.; Charette, A. B. *J. Am. Chem. Soc.* **2003**, *125*, 6360; (f) Buonora, P.; Olsen, J.-C.; Oh, T. *Tetrahedron* **2001**, *57*, 6099; (g) Kita, Y.; Maekawa, H.; Yamasaki, Y.; Nishiguchi, I. *Tetrahedron* **2001**, *57*, 2095; (h) Rutjes, F. P. J. T.; Tjen, K. C. M. F.; Wolf, L. B.; Karstens, W. F. J.; Schoemaker, H. E.; Hiemstra, H. *Org. Lett.* **1999**, *1*, 717; (i) Abell, A. D.; Gardiner, J.; Phillips, A. J.; Robinson, W. T. *Tetrahedron Lett.* **1998**, *39*, 9563; (j) Braña, M. F.; Morán, M.; Pérez de Vega, M. J.; Pita-Romero, I. *J. Org. Chem.* **1996**, *61*, 1369; (k) Norman, M. H.; Heathcock, C. H. *J. Org. Chem.* **1988**, *53*, 3370.
- For some examples on the synthesis of multiplysubstituted 1,2,3,4-tetrahydropyridines, see: (a) Caplan, J. F.; Sutherland, A.; Vederas, J. C. *J. Chem. Soc., Perkin Trans. 1* **2001**, 2217; (b) Harrison, T. J.; Dake, G. R. *J. Org. Chem.* **2005**, *70*, 10872; (c) Li, X.; Song, L.; Xing, C.; Zhao, J.; Zhu, S. *Tetrahedron* **2006**, *62*, 2255; (d) Han, R.-G.; Wang, Y.; Li, Y. Y.; Xu, P. F. *Adv. Synth. Catal.* **2008**, *350*, 1474; (e) Zannatta, N.; da Fernandes, L. S.; Nachtigall, F. M.; Ceolho, H. S. *Eur. J. Org. Chem.* **2009**, 1435; (f) Sridharan, V.; Maiti, S.; Menéndez, J. C. *Chem.—Eur. J.* **2009**, *15*, 4565; (g) Liu, W.-B.; Jiang, H.-F.; Zhu, S.-F.; Wang, W. *Tetrahedron* **2009**, *65*, 7985; (h) Sperger, C. A.; Mayer, P.; Wanner, K. T. *Tetrahedron* **2009**, *65*, 10463.
- (a) Zhang, M.; Jiang, H.; Liu, H.; Zhu, Q. *Org. Lett.* **2007**, *9*, 4111; (b) Cao, H.; Wang, X.; Jiang, H.; Zhu, Q.; Zhang, M.; Liu, H. *Chem.—Eur. J.* **2008**, *14*, 11623; (c) Zhu, Q.; Jiang, H.; Li, J.; Zhang, M.; Wang, X.; Qi, C. *Tetrahedron* **2009**, *65*, 4604.
- Cao, H.; Jiang, H.-F.; Qi, C.-R.; Yao, W.-J.; Chen, H.-J. *Tetrahedron Lett.* **2009**, *50*, 1209.
- Zhang, M.; Jiang, H. F.; Wang, A. Z. *Synlett* **2007**, *20*, 3214.
- Jiang, H.; Mai, R.; Cao, H.; Zhu, Q.; Liu, X. *Org. Biomol. Chem.* **2009**, *7*, 4943.
- Müller, T. E.; Hultsch, K. C.; Yus, M.; Foubelo, F.; Tada, M. *Chem. Rev.* **2008**, *108*, 3795.
- (a) Tömösközi, I.; Gruber, L.; Kovács, G.; Székely, I.; Simonidesz, V. *Tetrahedron Lett.* **1976**, *17*, 4639; (b) Snider, B. B.; Rodini, D. J.; Kirk, T. C.; Cordova, R. J. *Am. Chem. Soc.* **1982**, *104*, 555; (c) Tian, X.; Jaber, J. J.; Rychnovsky, S. D. *J. Org. Chem.* **2006**, *71*, 3176.
- For representative recent examples of proline-catalyzed reactions, see: (a) List, B.; Lerner, R. A.; Barbas, C. F., III. *J. Am. Chem. Soc.* **2000**, *122*, 2395; (b) Sakthivel, K.; Notz, W.; Bui, T.; Barbas, C. F., III. *J. Am. Chem. Soc.* **2001**, *123*, 5260; (c) Zhou, Y.; Shan, Z. *J. Org. Chem.* **2006**, *71*, 9510; (d) Davies, H. J.; Ruda, A. M.; Tomkinson, N. C. O. *Tetrahedron Lett.* **2007**, *48*, 1461; (e)

- Chandrasekhar, S.; Vijeender, K.; Sridhar, C. *Tetrahedron Lett.* **2007**, *48*, 4935; (f) Sengupta, S.; Duan, H.; Lu, W.; Petersen, J. L.; Shi, X. *Org. Lett.* **2008**, *10*, 1493; (g) Hahn, B. T.; Fröhlich, R.; Harms, K.; Glorius, F. *Angew. Chem., Int. Ed.* **2008**, *47*, 9985; (h) Zhong, C.; Chen, Y.; Petersen, J. L.; Akhmedov, N. G.; Shi, X. *Angew. Chem., Int. Ed.* **2009**, *48*, 1279; (i) Chandler, C.; Galzerano, P.; Michrowska, A.; List, B. *Angew. Chem., Int. Ed.* **2009**, *48*, 1978; (j) Mukhopadhyay, C.; Tapaswi, P. K.; Butcher, R. J. *Tetrahedron Lett.* **2010**, *51*, 1797.
13. (a) Lu, C.; Lu, X. *Tetrahedron* **2004**, *60*, 6575; (b) Li, Z.; Li, C.-J. *Eur. J. Org. Chem.* **2005**, 3173; (c) Ranu, B. C.; Banerjee, S. *Org. Lett.* **2005**, *7*, 3049; (d) Zhang, J.; Sarma, K. D.; Curran, T. T.; Belmont, D. T.; Davidson, J. G. *J. Org. Chem.* **2005**, *70*, 5890; (e) Zhang, Y.; Li, C.-J. *Angew. Chem., Int. Ed.* **2006**, *45*, 1949; (f) Basheer, A.; Rappoport, Z. *J. Org. Chem.* **2006**, *71*, 9743; (g) Ranu, B. C.; Chattopadhyay, K.; Adak, L. *Org. Lett.* **2007**, *9*, 4595; (h) Fotouhi, L.; Heravi, M. M.; Fatehi, A.; Bakhtiari, K. *Tetrahedron Lett.* **2007**, *48*, 5379; (i) Saeedi, M.; Heravi, M. M.; Beheshtiha, Y. S.; Oskooie, H. A. *Tetrahedron* **2010**, *66*, 5345.
14. FeCl₃-catalyzed formation of β -enamino carbonyl compounds has been reported already, see: Hebbache, H.; Hank, Z.; Boutamine, S.; Meklati, M.; Bruneau, C.; Renaud, J.-L. *C.R. Chim.* **2008**, *11*, 612.